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Ziegler, who was a [Patron Member](#) of the [SS](#)<sup>[9]</sup> received the [War Merit Cross](#) 2nd Class in October 1940.<sup>[10]</sup>

From 1943 until 1969, Ziegler was the Director of the [Max Planck Institute for Coal Research](#) (Max-Planck-Institut für Kohlenforschung) formerly known as the Kaiser-Wilhelm Institute for Coal Research (Kaiser-Wilhelm-Institut für Kohlenforschung) in Mülheim an der Ruhr as a successor to [Franz Fischer](#).<sup>[8]</sup>

Karl Ziegler was credited for much of the post war resurrection of chemical research in Germany and helped in founding the German Chemical Society (Gesellschaft Deutscher Chemiker) in 1949.

He served as president for five years.<sup>[7]([11]</sup> He was also the president of the German Society for Petroleum Science and Coal Chemistry (Deutsche Gesellschaft für Mineralölwissenschaft und Kohlechemie), which was from 1954 to 1957.<sup>[7]</sup> In 1971, The Royal Society, London, elected him as a Foreign Member.<sup>[12]</sup>

## Personal life [edit]

In 1922, Ziegler married Maria Kurtz.<sup>[2]</sup> They had two children, Erhart and Marianna.<sup>[3]</sup> His daughter, Dr. Marianna Ziegler Witte was a doctor of medicine and married a chief physical of a children's hospital (at that time) in the Ruhr. His son, Dr. Erhart Ziegler, became a physicist and patent attorney. In addition to his children, Karl Ziegler has five grandchildren by his daughter, and five by his son.<sup>[11]</sup> At least one of his grandchildren, Cordula Witte, attended his Nobel Prize reception as there is a picture of the two of them happily dancing.<sup>[5]</sup> Ziegler enjoyed traveling around the world with his family, especially on cruises. He even charted special cruises and airplanes for eclipse viewing. It was during a 1972 eclipse-viewing cruise with his grandson that Karl Ziegler became ill. He died a year later.<sup>[6]</sup>

Ziegler and his wife were great lovers of the arts, particularly paintings. Karl and Maria would present each other with paintings for birthdays, Christmases, and anniversaries. They amassed a large collection of paintings, not necessarily of one particular period, but of paintings they enjoyed. Maria, being an avid gardener, particularly enjoyed flower paintings by [Emil Nolde](#), [Erich Heckel](#), [Oskar Kokoschka](#), and [Karl Schmidt-Rottluff](#). Karl enjoyed pictures of the places that he and his wife called home, including pictures of Halle and the [Ruhr valley](#). Forty-two images from their shared collection were incorporated into a foundation, bequeathed to the Mülheim Ziegler Art Museum.<sup>[13]</sup>

As a man of many discoveries, Karl Ziegler was also a man of many patents. As a result of his patent agreement with the Max Planck Institute, Ziegler was a wealthy man. With part of this wealth, he set up the Ziegler Fund with some 40 million deutsche marks to support the institute's research.<sup>[6]</sup> Another namesake is the Karl-Ziegler-Schule, an urban high school that was founded on December 4, 1974, renaming a previously existing school. The school is located in Mülheim, Germany.<sup>[13]</sup>

Karl Ziegler died in [Mülheim](#), Germany August 12, 1973.

## Scientific advancements [edit]

Throughout his life, Ziegler was a zealous advocate for the necessary indivisibility of all kinds of research. Because of this, his scientific achievements range from the fundamental to the most practical, and his research spans a wide range of topics within the field of chemistry. As a young professor, Ziegler posed the question: what factors contribute to the [dissociation](#) of carbon-carbon bonds in substituted [ethane](#) derivatives? This question was to lead Ziegler on to a study of [free radicals](#), [organometallics](#), [ring compounds](#), and, finally, [polymerization processes](#).<sup>[4]</sup>

### Free radical compounds [edit]

While still a doctoral student at [University of Marburg](#), Ziegler published his first major article which showed how halochromic (R<sub>3</sub>C<sup>+</sup>Z<sup>-</sup>) salts could be made from carbinols. Previous work had left the impression that halochromic salts or free radicals (R3C•) required R to be [aromatic](#). He was encouraged to try to synthesize similarly substituted free radicals, and successfully prepared 1,2,4,5-tetraphenylallyl in 1923 and pentaphenylcyclopentadienyl in 1925. These two compounds were much more stable than previous tri-valent carbon free radicals, such as [triphenylmethyl](#). His interest in the stability of tri-valent carbon free-radical compounds brought him to publish the first of many publications in which he sought to identify the steric and electronic factors responsible for the dissociation of hexa-substituted ethane derivatives.<sup>[14]</sup>

### Many-membered ring compounds [edit]

Ziegler's work with many-membered ring compounds also utilized the reactive nature of alkali metal compounds. He used strong bases such as the lithium and sodium salts of amines, to accomplish the cyclization of long-chain [hydrocarbons](#) possessing terminal cyano groups. The initially formed ring compound was then converted to the desired macrocyclic ketone product. Ziegler's synthetic method, which included running reactions at high dilution to favor the [intramolecular](#) cyclization over competing intermolecular reactions, resulted in yields superior to those of existing procedures (Laylin): he was able to prepare large-ringed alicyclic ketones, C<sub>14</sub> to C<sub>33</sub>, in yields of 60–80%.<sup>[4]</sup> An outstanding instance of this synthesis was the preparation of [muscone](#), the odiferous principle of animal musk by [Leopold Ružička](#).<sup>[11]</sup> Ziegler and co-workers published the first of their series of papers on the preparation of large ring systems in 1933. For his work in this area and in free-radical chemistry he was awarded the Liebig Memorial Medal in 1935.<sup>[14]</sup>

### Organometallic compounds [edit]

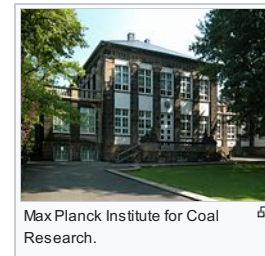
Ziegler's work with free radicals led him to the organo compounds of the [alkali metals](#). He discovered that ether scission opened a new method of preparing sodium and potassium alkyls,<sup>[11]</sup> and found that these compounds could easily be converted to the hexa-substituted ethane derivatives. The nature of the substituent could be easily and systematically altered using this synthetic route by simply changing the identity of the ether starting material.<sup>[14]</sup>

#### Lithium alkyls [edit]

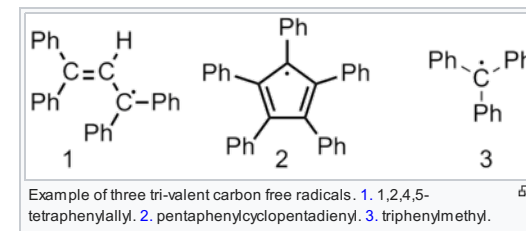
Later, in 1930, he directly synthesized lithium alkyls and aryls from metallic lithium and halogenated hydrocarbons. 4Li+2RX – 2RLi This convenient synthesis spurred numerous studies of RLi reagents by others, and now [organolithium reagents](#) are one of the most versatile and valuable tools of the synthetic organic chemist. Ziegler's own research on lithium alkyls and olefins was to lead directly to his discovery of a new polymerization technique some 20 years later.

#### Living polymerization [edit]

In 1927, he found that when the [olefin stilbene](#) was added to an ethyl ether solution of phenylisopropyl potassium, an abrupt color change from red to yellow took place. He had just observed the first addition of an organoalkali metal



Max Planck Institute for Coal Research.



Example of three tri-valent carbon free radicals. 1. 1,2,4,5-tetraphenylallyl. 2. pentaphenylcyclopentadienyl. 3. triphenylmethyl.

compound across a carbon-carbon double bond. Further work showed that he could successively add more and more of the olefinic hydrocarbon **butadiene** to a solution of phenylisopropyl potassium and obtain a long-chain hydrocarbon with the reactive organopotassium end still intact. **Oligomers** such as these were the forerunners of the so-called "**living polymers**"

## Polyethylene [edit]

*Main article: polyethylene*

Since Ziegler was working at the **Max Planck Institute for Coal Research**, **ethylene** was readily available as a byproduct from coal gas. Because of this cheap feedstock of ethylene and the relevance to the coal industry, Ziegler began experimenting with ethylene, and made it a goal to synthesize polyethylene of high molecular weight. His attempts were thwarted because a competing elimination reaction kept occurring causing an anomalous result: instead of ethylene being converted into a mixture of higher aluminum alkyls, its dimer, **1-butene**, was almost the only product. It was reasoned that a contaminant must have been present to cause this unexpected elimination reaction,<sup>[14]</sup> and the cause was eventually determined to be traces of nickel salts. Ziegler realized the significance of this finding; if a nickel salt could have such a dramatic influence on the course of an ethylene-aluminum alkyl reaction, then perhaps another metal might delay the **elimination reaction**. Ziegler and his student H. Breil found that salts of **chromium**, **zirconium**, and especially **titanium** did not promote the R2AlH-elimination but, instead, enormously accelerated the "growth" reaction. Simply passing ethylene, at atmospheric pressure, into a catalytic amount of TiCl3 and Et2AlCl dissolved in a higher alkane led to the prompt deposition of polyethylene. Ziegler was able to obtain **high molecular weight polyethylene** (MW > 30,000) and, most importantly, to do so at low ethylene pressures. The Ziegler group suddenly had a polymerization procedure for ethylene superior to all existing processes.

## Ziegler–Natta catalyst [edit]

*Main article: Ziegler–Natta catalyst*

In 1952, Ziegler disclosed his **catalyst** to the Montecatini Company in Italy, for which **Giulio Natta** was acting as a consultant. Natta denoted this class of catalysts as "Ziegler catalysts" and became extremely interested in their ability and potential to stereoregularly polymerize  $\alpha$ -olefins such as propene.<sup>[14]</sup> Ziegler, meanwhile concentrated mainly on the large-scale production of **polyethylene** and copolymers of **ethylene** and **propylene**. Soon the scientific community was informed of his discovery. Highly **crystalline** and stereoregular polymers that previously could not be prepared became synthetically feasible. For their work on the controlled polymerization of hydrocarbons through the use of these novel organometallic catalysts, Karl Ziegler and Giulio Natta shared the 1963 Nobel Prize in Chemistry.

## Awards and honours [edit]

Karl Ziegler received many awards and honors. The following highlights some of the most significant awards:

- **Liebig-Denk Münze medal** (1935); This medal was awarded by the present day German Chemical Society and is given to a German chemist for their outstanding accomplishments and creativity. Ziegler received the award because of his work in the synthesis of multi-membered ring systems and stable tri-valent carbon radicals.
- **War Merit Cross**, 2nd class (19 October 1940)
- Carl Duisberg Plakette (1953); This award is given for outstanding service to the promotion of chemistry from the German Chemical Society.
- **Lavoisier Medal** (1955); This award is given by the French Chemical Society to scientists in various disciplines of chemistry.
- Carl Engler Medal (1958); This award is given by the German Society of Petroleum Science and Coal Chemicals, of which he was the president in earlier years.
- **Werner von Siemens Ring** (1960); This ring is awarded by the **Werner von Siemens Foundation** and is considered the highest German award for individuals who by their performance and skills opened up new technological paths.
- **Nobel Prize in Chemistry** (1963); "[His] excellent work on organometallic compounds has unexpectedly led to new polymerization reactions and thus paved the way for new and highly useful industrial processes."<sup>[7]</sup>
- **Swinburne Medal** of the **Plastics Institute**, London (1964); This award recognizes an individual who has made a significant advancement to the science, engineering or technology of plastics.
- **Grand Merit Cross with Star and Sash** (*Großes Verdienstkreuz mit Stern und Schulterband*) of the Federal Republic of Germany (1964)
- International Synthetic Rubber Medal of Rubber and Plastics Age (1967)
- Grand **Federal Cross of Merit** (Order of Merit of the Federal Republic of Germany)(1969); Ziegler was given this award for his work in the area of Science and Technology.
- **Pour le Mérite for Arts and Sciences** (formerly Peace Class) (1969)
- Foreign Member of the **Royal Society** (1971)<sup>[15]</sup>
- **Wilhelm Exner Medal** (1971).<sup>[16]</sup>
- Memorial tablet of the **German Chemical Society** under the Historic Landmarks of Chemistry program (*Historische Stätten der Chemie*) at the **Max Planck Institute for Coal Research** in **Mülheim an der Ruhr** (2008)
- Honorary doctorates from the **Technical University of Hannover**, **Giessen University**, **University of Heidelberg** and **Darmstadt Technical University**
- The *Karl-Ziegler-Schule* in Mülheim was named after Ziegler
- The Karl Ziegler Foundation is located at the German Chemical Society and gives the Science Award, the Karl Ziegler Prize (worth 50,000 euros)<sup>[citation needed]</sup>

## See also [edit]

- **List of Nobel laureates in Chemistry**

## References [edit]

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Memorial tablet of the GDCh.

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## External links [edit]

- Nobel biography

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